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**AMENDMENTS TO THE CLAIMS** 

This listing of claims replaces all prior versions of listing of claims, and listing of claims

in the application.

<u>Listing of Claims</u>

1. (Currently Amended) A liquid ionic compound comprising a cation which is a

complex of a neutral ligand selected from the group consisting of organic substituted and

unsubstituted alkyl amines and crown ethers with a metal ion selected from the group

consisting of Na<sup>+</sup>, K<sup>+</sup>, Li<sup>+</sup>, Ca<sup>2+</sup>, Ag<sup>+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup>, Cd<sup>2+</sup>, Ni<sup>2+</sup>, Hg<sup>2+</sup>, Co<sup>3+</sup> and Fe<sup>3+</sup> and an

anion which is a conjugate anion of the metal ion, wherein said anion comprises sulfur or

phosphorous said cation and said anion comprising a substantially new liquid.

2. (Previously Presented)

An ionic compound according to claim 1 which is a

liquid below 100°C.

3. (Previously Presented)

An ionic compound according to claim 2 which is a

liquid at room temperature.

4. (Previously Presented) An ionic compound according to claim 1 which is

electrically conductive in the absence of a solvent.

5. (Previously Presented)

An ionic liquid according to claim 1 which is

hydrophobic.

6. (Previously Presented) An ionic compound according to claim 1 wherein

said neutral organic ligand is a crown ether.

7. (Cancelled)

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- 8. (Previously Presented) An ionic compound according to claim 1 wherein said conjugate anion is bis(trifluoromethane)sulfonimide, boron trifluoride, nitrate, sulfate, phosphate, hexafluorophosphate and dicyanamide.
- 9. (Currently Amended) A method for forming a substantially neat ionic liquid ionic compound, the method comprising mixing a neutral ligand selected from the group consisting of organic substituted and unsubstituted alkyl amines and crown ethers with a metal ion selected from the group consisting of Na<sup>+</sup>, K<sup>+</sup>, Li<sup>+</sup>, Ca<sup>2+</sup>, Ag<sup>+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup>, Cd<sup>2+</sup>, Ni<sup>2+</sup>, Hg<sup>2+</sup>, Co<sup>3+</sup> and Fe<sup>3+</sup> and with the salt of a metal cation and its conjugate anion at room temperature.
- 10. (Original) A method according to claim 9 wherein said neutral organic ligand is a crown ether.
- 11. (Original) A method according to claim 10 wherein the metal cation is selected from the group consisting of sodium, potassium, lithium and calcium.
- 12. (Cancelled)
- 13. (Original) A method according to claim 12 wherein said metal cation is selected from the group consisting of silver, zinc, copper, cadmium, nickel, mercury and iron.
- 14. (Previously Presented) A method according to claim 9 wherein said conjugate anion is bis(trifluoromethane)sulfonimide, boron trifluoride, nitrate, sulfate, phosphate, hexafluorophosphate and dicyanamide.
- 15.-20. (Cancelled)